

EFFICIENT SIMULATION OF REACTIVE FLOW IN RESERVOIRS ROCKS AT THE PORE SCALE

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1

Introduction

Model workflow

2

pH-based reaction model

Dissolution patterns in a Carbonate Rock

3

Constant reaction-rate model

Carbonate Rock and Sandstone pore cements

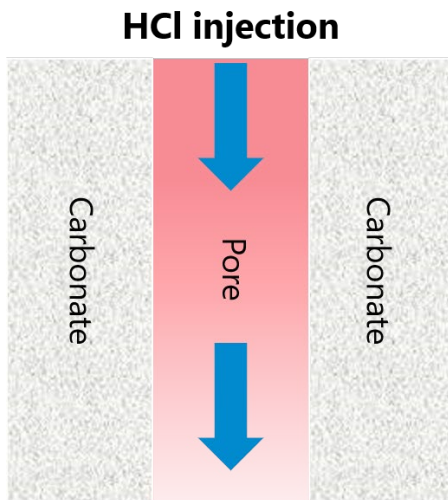
4

Conclusion and Outlook

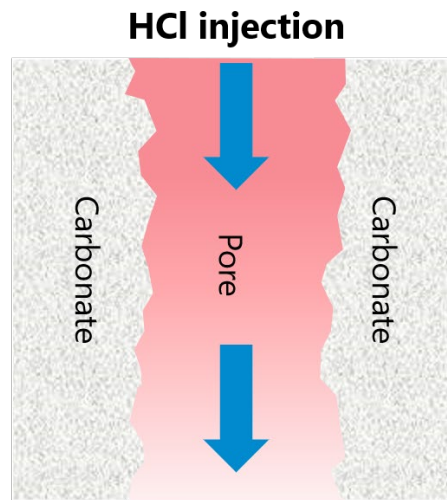
Reactive flow with complete aqueous geochemistry

Acidizing treatments in carbonate reservoirs

- HCl injection into carbonate
- Effect: Dissolution, enlargement of pore space
- Establish a higher permeability
- Keep mechanical stability



Initial stage



Dissolved stage

Numerical parameter studies

- Efficient optimization of process parameters

*e.g., acid concentration,
injection velocity, and
pore-scale structure*



**Such parameter studies
require efficient solvers**

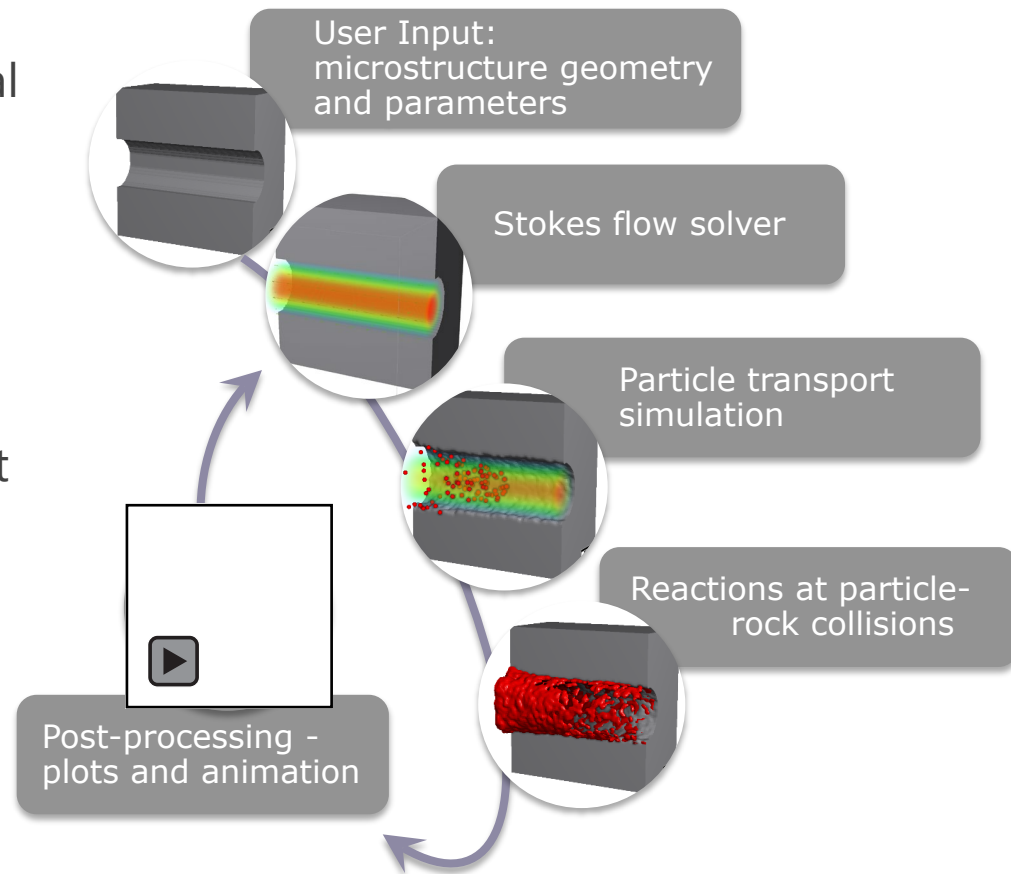
Our Motivation: Efficient model
capable of performing digital
reactive flow experiments in
large pore-scale geometries.

Our **Reactive Flow** feature computes dissolution and precipitation of mineral phases during continuous inflow of reactants (e.g., acid) and predicts:

- 4D rock alteration
- Dissolution regime
- Porosity-permeability development

Reactive Flow applications in DRP:

- Permeability enhancement
e.g., in carbonate reservoirs
- Permeability reduction
e.g., by salt precipitation



FLOW COMPUTATION

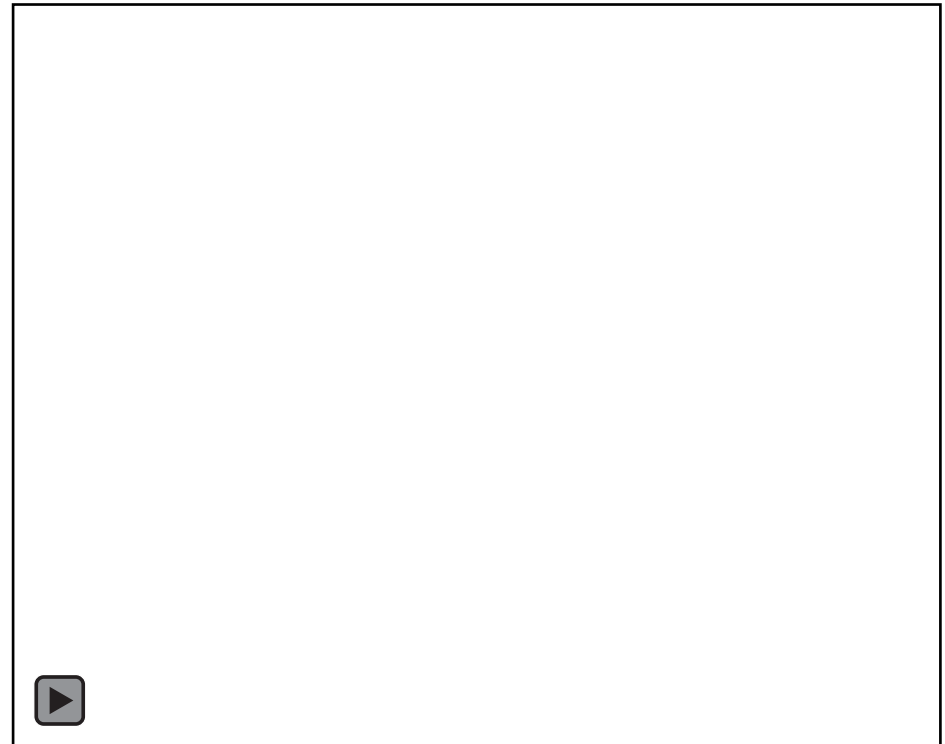
Advective motion is optional

- Flow field is computed in the current batch's structure
LIR Stokes solver of Linden et al. (2014)
- Flow is solved for a given fluid velocity
Every X time steps
- Boundary Conditions (BCs) may be set individually
*Periodic or symmetric BCs,
Convergence, Inlet/Outlet regions*



PARTICLE SIMULATION

- Compute particle movement
 - Advective motion (Streamlines)
 - Diffusive motion (Brownian motion)
 - Motion according to molecules



Reaction Rate & pH-based geochemical Models:

*We keep track of the collision points with the rock interface, where particles will react.
Particles' motion stops when they are geochemically exhausted (user-defined).*

REACTION – MODELS

Model I: pH-based model

At particle-rock collisions, a partial acid volume dissolves the reactive solid surface partially

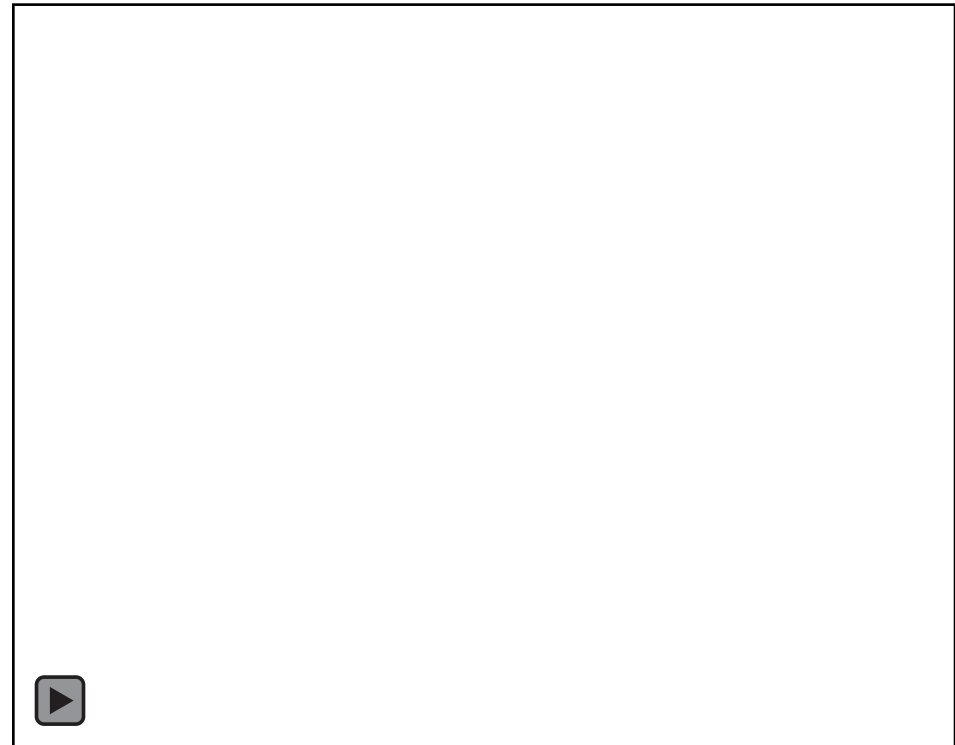
- Each particle represents an acid volume indicated via pH value

Model II: Reaction Rate Model

At particle-rock collisions, the solid surface reacts partially.

- Reaction is based on reaction rates, which are:
 - (1) *user-defined,*
 - (2) *mineral-specific, and*
 - (3) *global reaction rates*

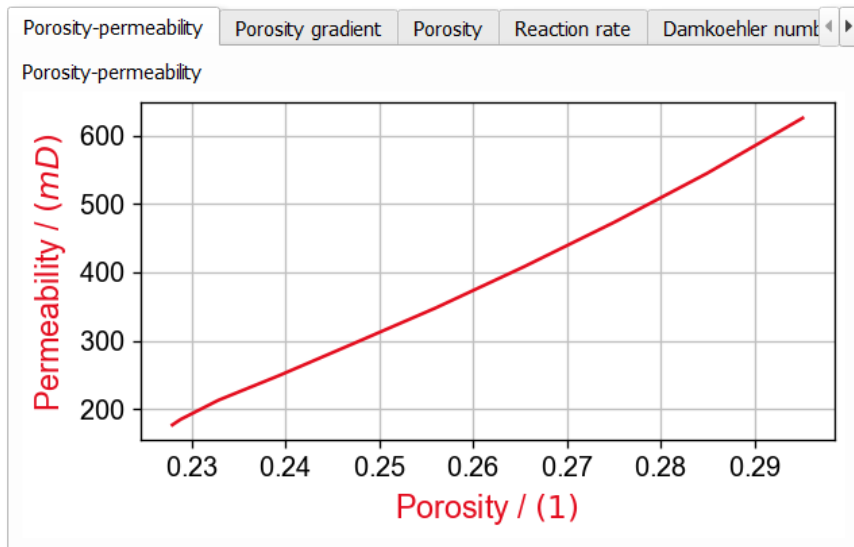
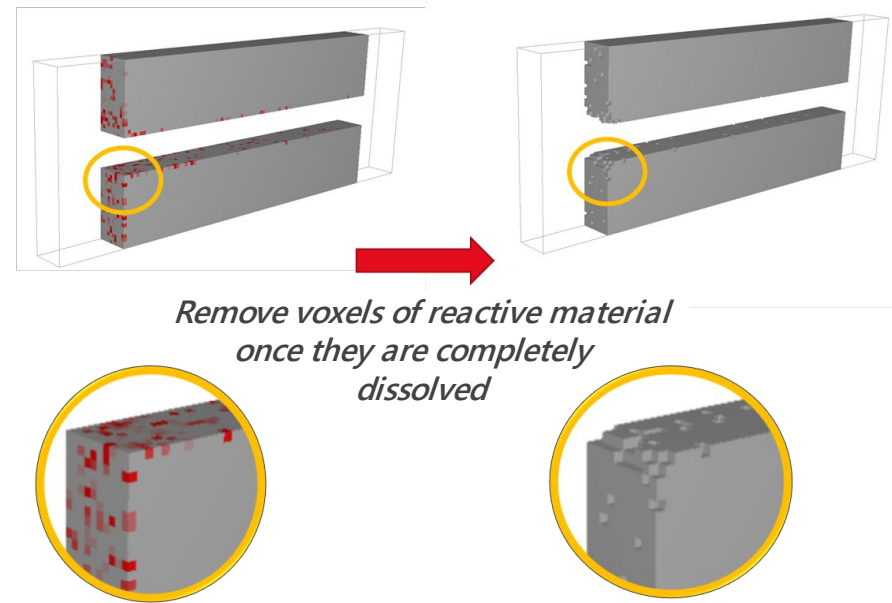
We keep track of the consumed acid volume of particles and of the solid fractions of reactive minerals.



is highlighted in red color.

DATA PROCESSING

- Rock structure is automatically updated
Remove dissolved voxels, precipitate voxels
- All data may be stored
and used for further analysis:
Mechanical stability (Bulk Modulus),
Conductivity, and more



- Automated post-processing
Porosity-Permeability relationship,
Reaction rate, Damköhler number,
3D Animations, amongst others

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GROSMONT CARBONATE SAMPLE

GROSMONT FORMATION, ALBERTA, CANADA

GEODICT

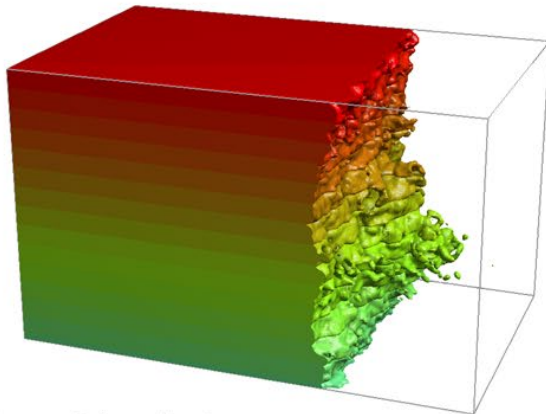
- Dimensions: 1024x1024x1024 voxel
- Resolution: 2.02 μm
- Porosity: 21 %
- Permeability range: 150 mD – 470 mD
→ heterogenous pore space
- Data set is published in DRP
benchmark paper (Andrä et al., 2013)



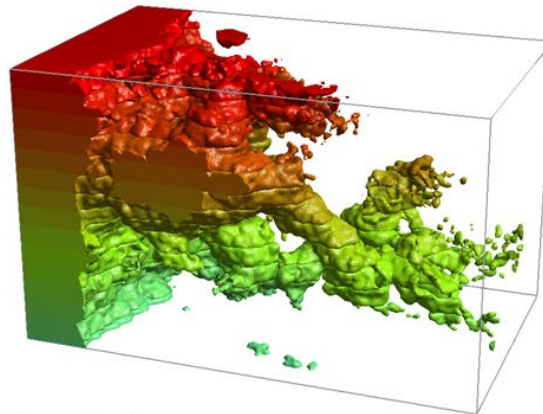
DISSOLUTION REGIMES

Goal: Replicate characteristic dissolution patterns

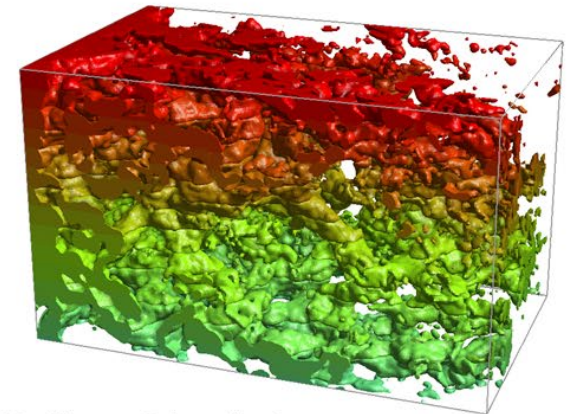
- **Face Dissolution:** Slow flow in relation to reaction rate
- **Wormhole:** Comparable transport rate and reaction rate
- **Uniform Dissolution:** Fast flow in relation to reaction rate



Face Dissolution



Wormhole



Uniform Dissolution

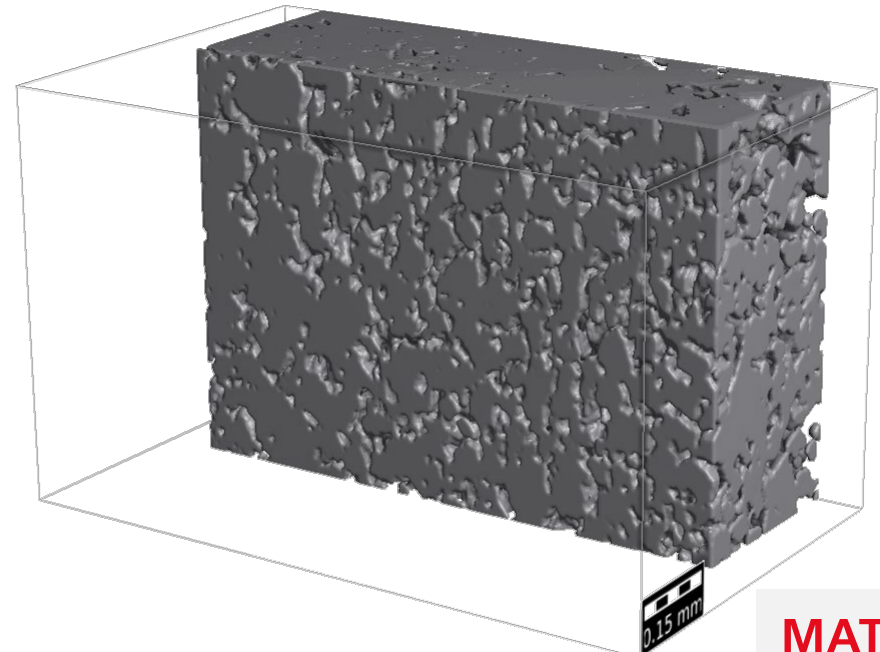
Goal: Replicate characteristic dissolution patterns

pH-based models		$v \left[\frac{m}{s} \right]$	$t_{sim} [s]$	pH	$\Delta\Phi [\%]$	$t_{runtime} [h]$	$\Delta permeability [mD]$
Face Dissolution	256x256x362	0.00001	500,000	3.0	11.5	30.6	160
Wormhole		0.0005	6,000	3.0	9.0	2.4	3,165
Uniform Dissolution		0.05	200	3.0	7.4	3.0	1,354

Grosmont carbonate

Andrä et al. (2013)

- Porosity $\Phi = 21.95 \%$
- Voxel length: $2.02 \mu m$
- Domain: $256 \times 256 \times 362$ voxel

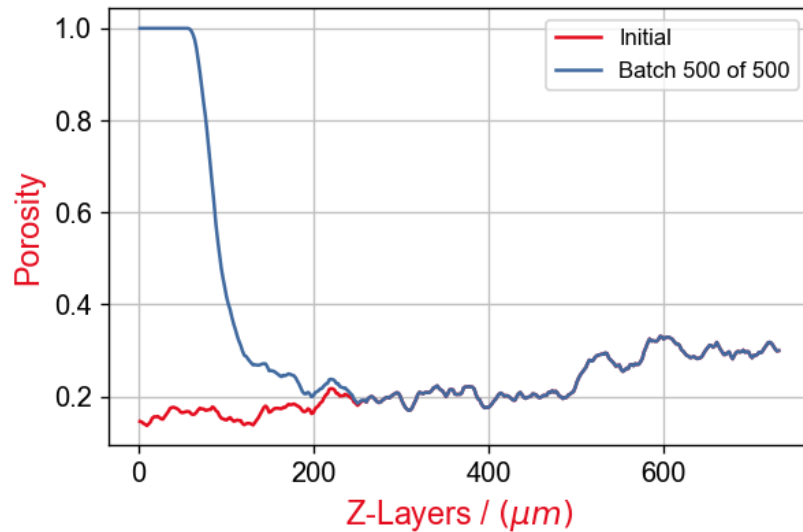


FACE DISSOLUTION

GROSMONT CARBONATE SAMPLE

Material Information:
ID 00: Water [invis.]
ID 01: Calcite

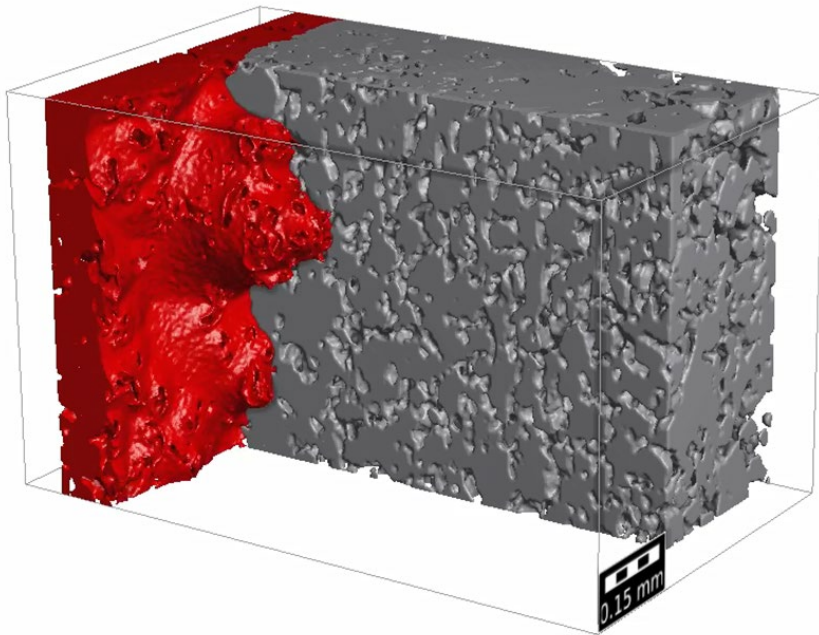
*Grosmont carbonate
Andrä et al. (2013)*



COMPARISON FACE DISSOLUTION PATTERN

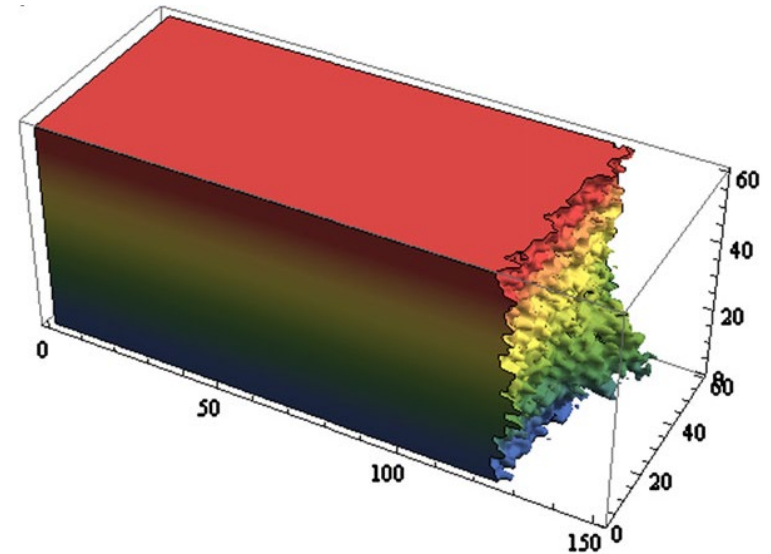
GEODict

GROSMONT CARBONATE SAMPLE



GeoDict Simulation

Pore scale modeling



Maheshwari et al. 2013

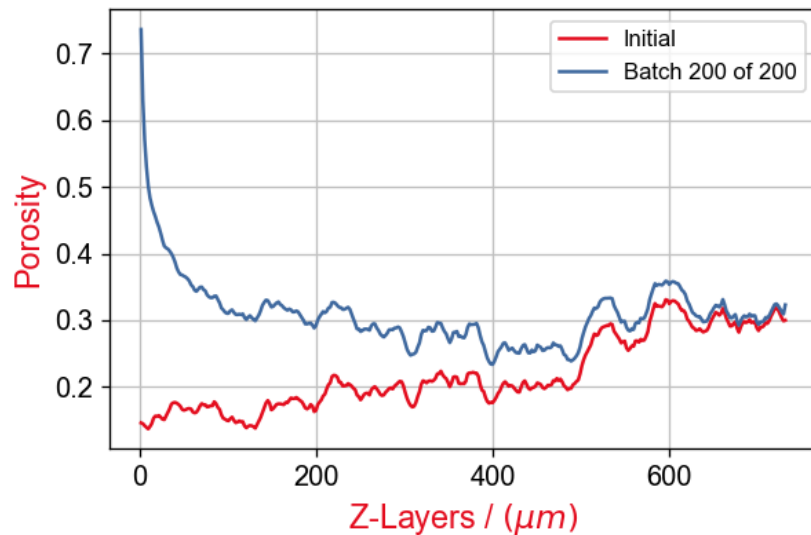
Continuum scale modeling

WORMHOLING

GROSMONT CARBONATE SAMPLE

Material Information:
ID 00: Water [invis.]
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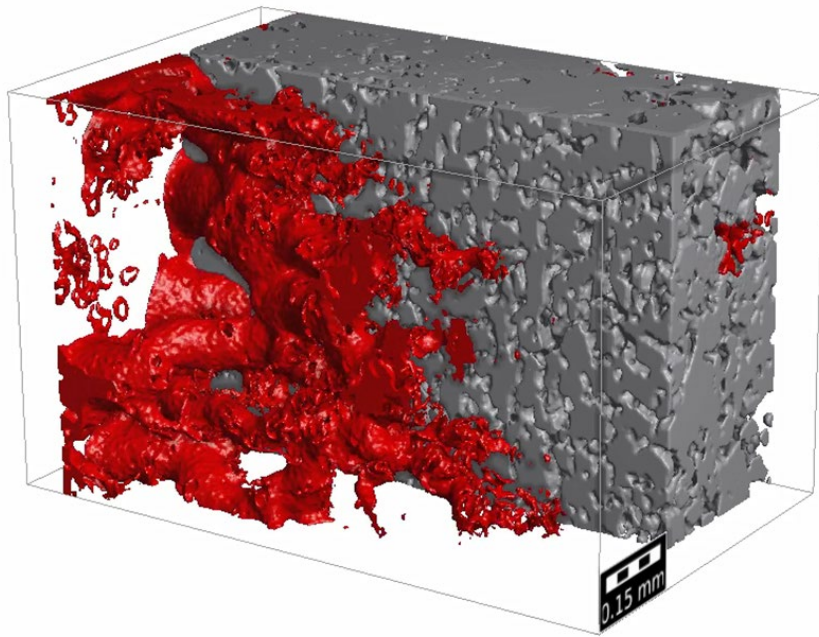
Grosmont carbonate
Andrä et al. (2013)



COMPARISON WORMHOLE PATTERN

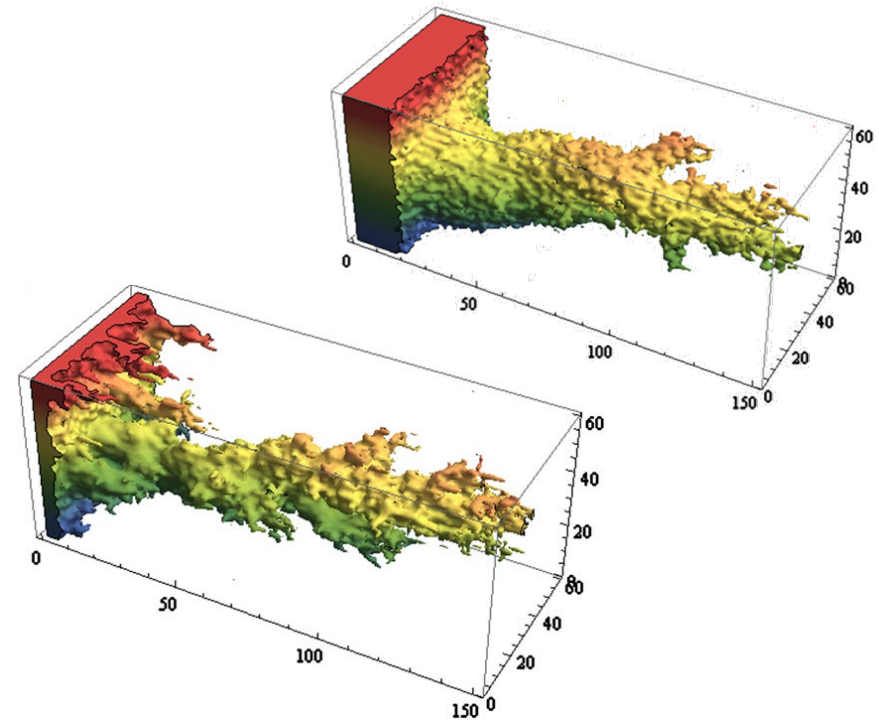
GROSMONT CARBONATE SAMPLE

GEODict



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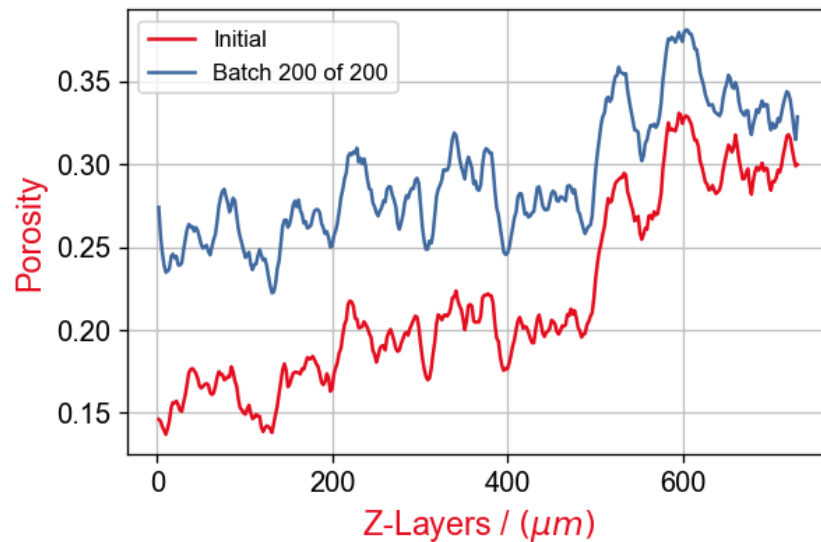
UNIFORM DISSOLUTION

GROSMONT CARBONATE SAMPLE

Material Information:

ID 00: Water [invis.]
ID 01: Calcite

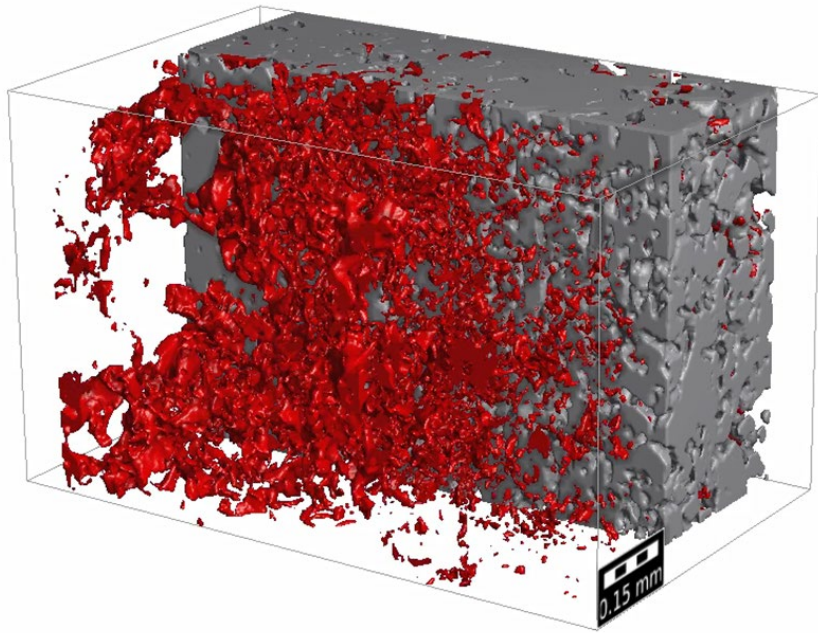
*Grosmont carbonate
Andrä et al. (2013)*



COMPARISON UNIFORM DISSOLUTION PATTERN

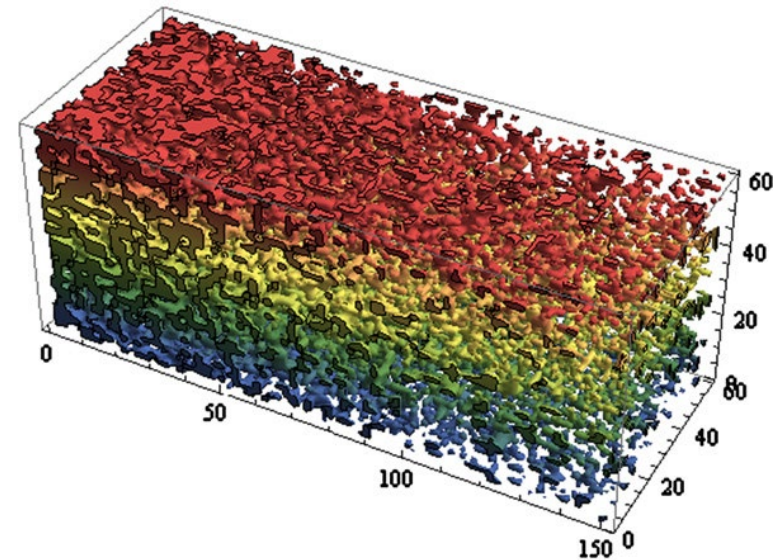
GEODict

GROSMONT CARBONATE SAMPLE



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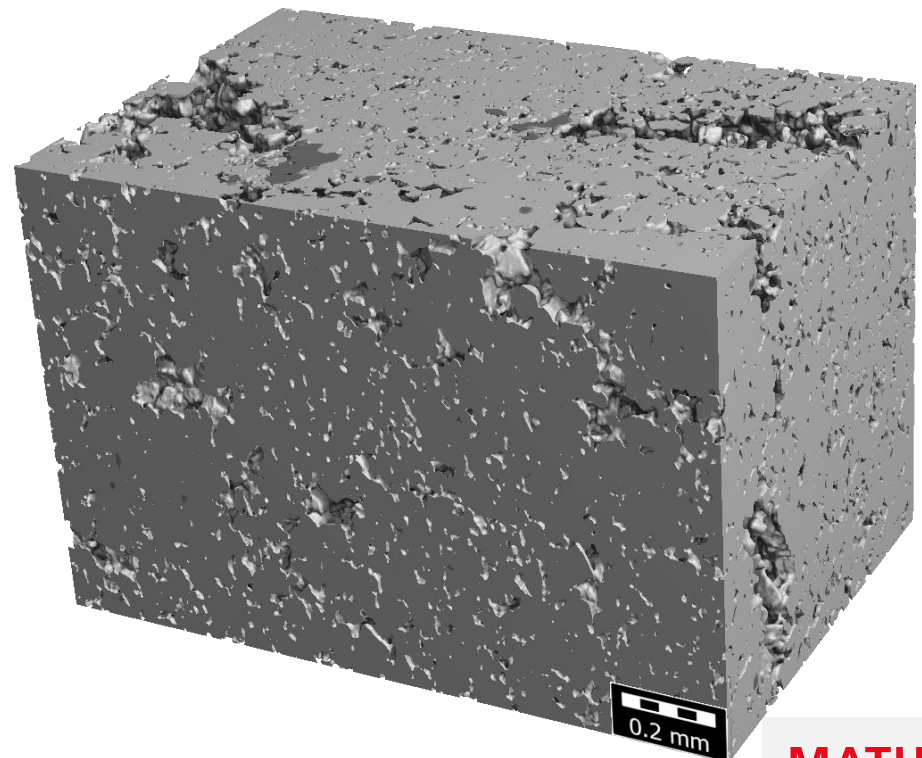
Goal: Perform kinetically-controlled carbonate dissolution

Reaction rate model		$v \left[\frac{m}{s}\right]$	$t_{\text{sim}} [a]$	$rate_{\text{Calcite}} \left[\frac{mol}{m^2s}\right]$	$\Delta\Phi [\%]$	$t_{\text{runtime}} [h]$
Carbonate Dissolution	512x512x768	0.00005	20	0.0001	10.98	28.8

Grosmont carbonate

Andrä et al. (2013)

- Porosity $\Phi = 23.53 \%$
- Voxel length: $2.02 \mu m$
- Domain: $512 \times 512 \times 768$ voxel

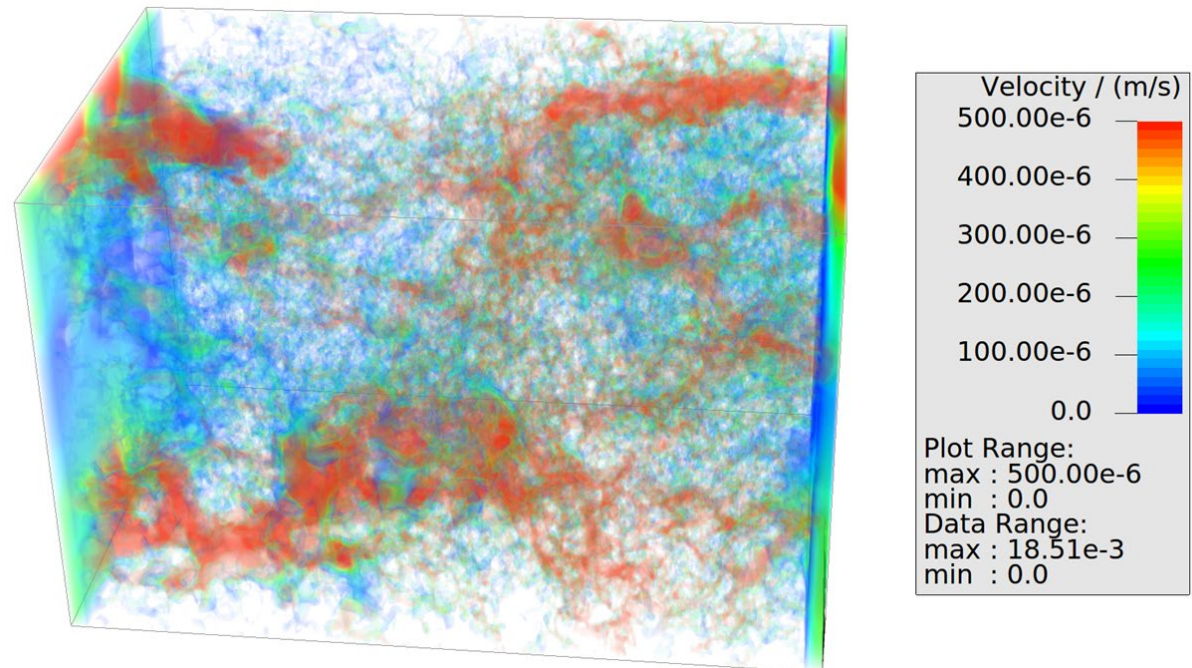


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Grosmont carbonate
Andrä et al. (2013)

- Initial flow field
 - Heterogeneous
 - Preferred flow path



GROSMONT CARBONATE SAMPLE

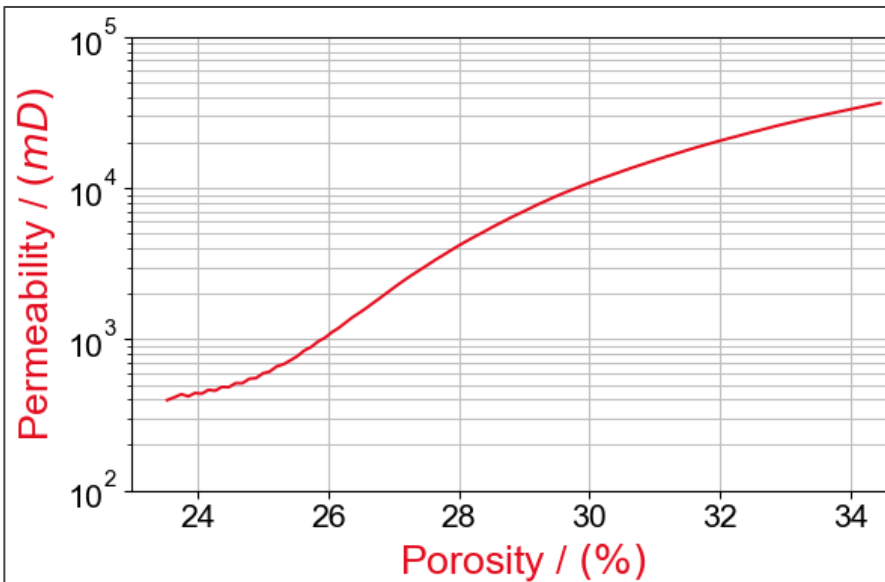
KINETIC CARBONATE DISSOLUTION

GEODICT

Parameter	Value
$t_{\text{simulated}}$	20 a
t_{runtime}	28.8 h

Material Information:
ID 00: Water [invis.]
ID 01: Calcite
ID 02: Inclusions
ID 03: Inclusions

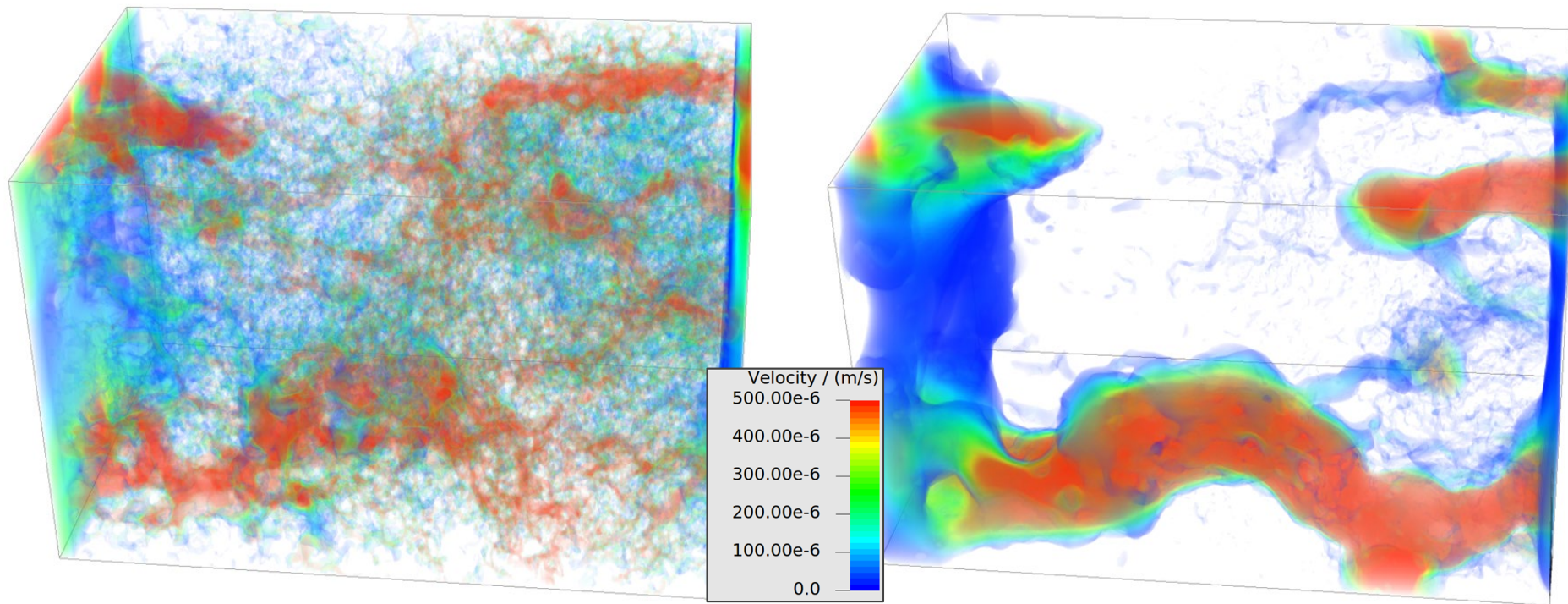
Grosmont carbonate



GROSMONT CARBONATE SAMPLE

KINETIC CARBONATE DISSOLUTION

GEODICT



Initial flow field

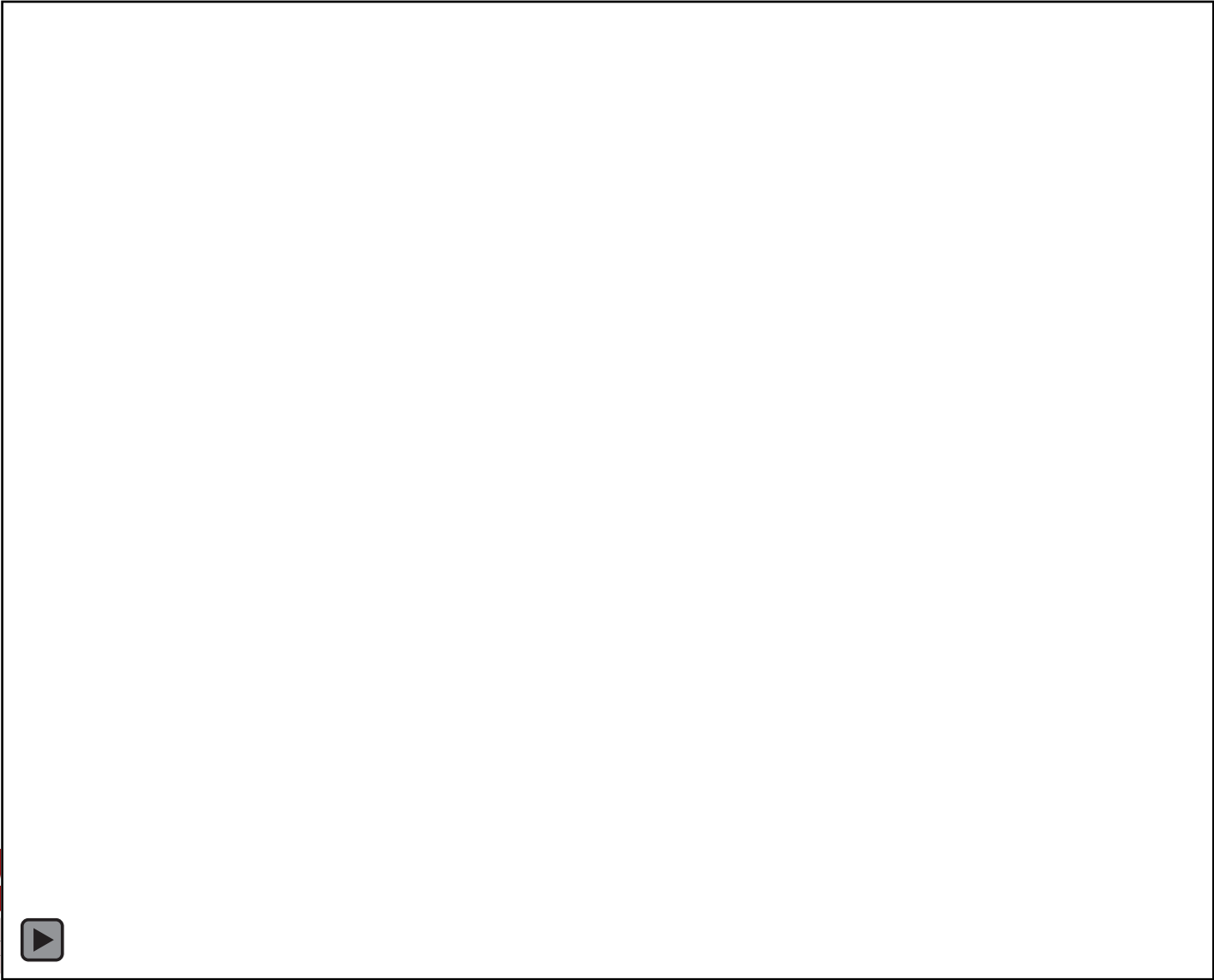
Final flow field

Grosmont carbonate
Andrä et al. (2013)

Channeling Dissolution Pattern

BEREA SANDSTONE SAMPLE

SELECTIVE DISSOLUTION – 2 PHASES



Information:
- color [invis.]
- quartz
- calcite
- spar
- on
- solved

Structure (v

720x720x



a sandstone
et al. (2013)

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- We presented a numerical model to simulate reactive flow
 - Different geochemical approaches
- We evaluated the model by reproducing characteristic dissolution patterns
- Computations can be run on a state-of-the-art desktop workstation
- Computational domains of $> 1000^3$ voxels
- On going advances in performing Digital Reactive Flow Experiments

Andrae, H., Combaret, N., Dvorkin, J., Glatt, E., Junehee, H., Kabel, M., Keehm, Y., Krzikalla, F., Lee, M., Madonna, C., Marsh, M., Mukerji, T., Saenger, E., Sain, R., Saxena, N., Ricker, S., Wiegmann, A., Zhan, A., "Digital rock physics benchmarks Part I: Imaging and segmentation", Computers & Geosciences, 43, 25-32, 2013.

Fischer, C., Arvidson, R.S., Lüttge, A., How predictable are dissolution rates of crystalline material? Geochim. Cosmochim. Acta 98, 177-185, 2012.

Maheshwari, P., Ratnakar, R.R., Kalia, N. and Balakotaiah, V., 3-D simulation and analysis of reactive dissolution and wormhole formation in carbonate rocks. Chemical Engineering Science, 90, 258-274, 2013.

Menke, H., Andrew, M.G., Blunt, M.J., Bijeljic, B., Reservoir condition imaging of reactive transport in heterogeneous carbonates using fast synchrotron tomography – Effect of initial pore structure and flow conditions. Chemical Geology 428, 15–26, 2016.

Steefel, C.I., Appelo, C.A.J., Arora, B., Jacques, D., Kalbacher, T., Kolditz, O., Lagneau, V., Lichtner, P.C., Mayer, K.U., Meeussen, J.C.L., Molins, S., Moulton, D., Shao, H., Šimůnek, J., Spycher, N., Yabusaki, S.B., Yeh, G.T., Reactive transport codes for subsurface environmental simulation. Computational Geosciences 19 (3), 445-478, 2015.

REACTIVE FLOW ADVANCEMENTS

- Simulate Mineral Dissolution & Crystallization upon consideration of the complete aqueous geochemistry
- Flow and transport solvers coupled to IPhreeqC (USGS)

Example: Reactive Flow setup for Carbonate dissolution using geochemical calcite reaction kinetics



pH value distribution changes continuously due to ongoing dissolution reactions in a Grosmont carbonate sample of Andrä et al. (2013).



Ca²⁺ concentration increases in the pore space due to continuous dissolution reactions in a Grosmont carbonate sample of Andrä et al. (2013).

THANK YOU FOR YOUR ATTENTION

GEODict

